## SYNTHESIS OF (±)DURYNE<sup>+</sup> V.H. Deshpande<sup>\*</sup>, B.K. Upadhye and R.D. Wakharkar National Chemical Laboratory, Pune 411 008, India

Summary: Synthesis of cytotoxic metabolite, duryne (1), involving symmetrical synthetic strategy is described.

There has been considerable growth in the newly discovered biologically active marine natural products in the last decade<sup>1</sup>. Recently, Wright et al.<sup>2</sup> reported the isolation of duryne (I) from marine sponge Cribrochaling dura which inhibits the growth of a number of in vitro tumour cell lines. The structure of duryne was established on the basis of spectroscopic analysis as a symmetrical C30 diol, consisting of two terminal acetylenic groups, two asymmetric centres and three olefinic bonds. Although the geometry of two double bonds at  $C^4$  and  $C^{4'}$ , is E, the geometry of central double bond  $(-HC^{15}-CH^{15'})$  and absolute configuration remain to be established. Prompted by its symmetry, recent origin, geometrical uncertainty of the central double bond, biological

activity and scarce availability via a tedious process of extraction from marine sources, we initiated a research programme which secures an expeditious entry into duryne frame. Based on the use of dialkylated acetylene to obtain both isomers E and Z of  $C^{15}$ - $C^{15'}$  double bond, herein we report the first synthesis of **Ia** and **Ib** having E and Z geometry of the central  $C^{15}-C^{15'}$  double bond respectively (Scheme-I).

SCHEME - I  
HC=CH
$$\rightarrow$$
XC = CY $-\frac{|||-V||, V, V|||}{P}$ R $-(CH_2)_9$ -CH=CH-(CH\_2)\_9-R

		3: X=H, Y=(CH <sub>2</sub> ) <sub>10</sub> -OTHP	5: R=CH <sub>2</sub> OTH	Р
Reage	ent se	4: X=Y ≈(CH <sub>2</sub> ) <sub>10</sub> —OTHP	6: R=CH <sub>2</sub> OH	
			7:R=CHO	
i.	n-BuLi, Br(CH <sub>2</sub> ) <sub>10</sub> -OTHP( <u>2)</u> , THF, HMPA;		8; R=HC =CI	
ii.	n-BuLi, <u>2</u> , HMPA;		8: R=HC ==CI	H — COOEt
iiia.	LAH, Diglyme, 140° Pd/CaCO <sub>3</sub> , H <sub>2</sub> ,Quinoline (cat.); MeOH, conc. HCl (cat.); PCC, CH <sub>2</sub> Cl <sub>2</sub> ;		9: R=HC=CH	-CH20H
iiib.			2 10:R=HC=CHCHO	
iv.				
v.			l:R=HC=CHCH-C≡CH	
vi.	Ph <sub>3</sub> P = CHCOOEt, E	Benzene;		1
vii.	DIBAL-H, Toluene,	-78ºC;		OH
viii.	HC ≡CMgBr, THF	_	a = E isomer	b = Z isomer

1,10-Decanediol was monobrominated $^3$  with aq. HBr (48%) followed by protection of hydroxyl group as THP-ether [DHP, conc.HCl (cat.)] to give THP-ether of ID-bromodecanol 2 (76%). Lithium acetvlide<sup>4</sup> (prepared from acetylene and n-BuLi in THF at O°C) was alkylated with **2** in HMPA

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to obtain monoalkylated acetylene  $3^5$  in 90% yield. It was again alkylated<sup>6</sup> with 2 using n-BuLi in HMPA to give dialkylated acetylene derivative  $4^7$  (60%) which was partially reduced<sup>8</sup> (LAH in diglyme, 140°C, 48 h) to obtain E olefin 5a (80%). The dialkylated acetylene derivative 4 was partially reduced [10% Pd on CaCO<sub>3</sub> in n-hexane,  $H_2$ , quinoline (cat.)] to obtain Z olefin 5b (95%). The diols 6a (m.p. 78-79°C) and **6b** (m.p. 61-62°C) readily prepared in 98 and 96% yield respectively by deprotection of **5a** and **5b** [MeOH, conc.HCl (cat.)], were oxidised<sup>10</sup> (PCC, CH<sub>2</sub>Cl<sub>2</sub>, 3 h) to dialdehyde **7a** (60%, m.p. 55-56°C) and 7b (60%, m.p. 30-31°C) respectively. Subsequent Wittig olefination of 7a and 7b with triphenylcarbethoxymethylene phosphorane in refluxing benzene (20 h) furnished E olefinic diester 8a (98%) and 8b (99%) respectively. The diesters 8a and 8b on reduction<sup>12</sup> (DIBAL-H in toluene. -78°C) gave diols 9a (m.p. 61-62°C, 96%) and 9b (m.p. 52-54°C, 94%) respectively which on subsequent oxidation (PCC,  $CH_2CI_2$ , 2 h) furnished 10a (60%) and 10b (64%) respectively. Finally ethynylation<sup>13</sup> of 10a and 10b (H-CZC-MgBr, THF, room temperature) gave duryne la in 60% yield m.p. 38-40°C and and lb in 63% yield m.p. 32-33°C. The spectral data of la and lb coincided<sup>14</sup> well with the reported values of natural duryne, however, the geometry of the double bond at  $C^{15}-C^{15}$  in the natural duryne could not be established.

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## **References and Notes**

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